THE GENERATION OF VINYL CATIONS BY BECKMANN FRAGMENTATION $\text{OF } \alpha,\beta\text{--}\text{UNSATURATED KETOXIME DERIVATIVES}$

C.A. Grob and P. Wenk

Institute of Organic Chemistry, University of Basel, CH-4056
(Received in UK 2 September 1976; accepted for publication 27 September 1976)

As shown in the preceding communication 1 derivatives of acyclic anti-vinyl-methylketoximes $\underline{1}$ (e.g. $X = p-CH_3C_6H_4SO_3$) bearing electron releasing substituents R^2 and R^3 readily undergo π -3 assisted Beckmann rearrangement to enamides $\underline{3}$ by way of the azirine cation $\underline{2}$. This mechanism requires that the planes of the C=C and C=N groups intersect at an angle of ca. 90° during activation. Fragmentation to a nitrile and a vinyl cation $\underline{4}$ (route F) was not detected even when the latter should be relatively stable, as when $R^1 = p-(CH_3) \, _2NC_6H_4$.

On this basis fragmentation would be expected to occur if π -3 participation were suppressed for structural reasons. This has now been confirmed by the following study of the rates and products of the p-toluenesulfonates (tosylates) of 3-methyl-2-aryl-2-cyclopenten-1-one oximes 5a, 5b and 5c, the C=C and C=N groups of which are practically coplanar.

$$H_{3}C$$
 $H_{3}C$
 H

In 80 % ethanol at 110° the p-unsubstituted tosylate <u>5a</u> reacted slowly affording the apparent hydrolysis product, i.e. the cyclic ketone <u>6a</u>, in 83 % yield beside tarry material. Under the same conditions the p-CH₃O-derivative <u>5b</u> yielded ll % of fragmentation products, i.e. the Z- and E-enol ethers <u>7b</u> beside their hydrolysis product <u>8b</u>. In addition, 61 % of the cyclic ketone <u>6b</u> beside tar were formed. At 80° the more reactive p-(CH₃)₂N-derivative <u>5c</u> afforded 63 % of fragmentation products, i.e. equal amounts of the Z- and E-enol ethers <u>7c</u> and their hydrolysis product <u>8c</u> beside 10 % of the cyclic ketone <u>6c</u> and tar. As expected, no rearranged lactams <u>9</u> were formed. The configuration of the enol ethers <u>7</u> follows from their NMR-spectra, the singlet due to the CH₃C=C group appearing at lower field in the E-isomer.

Table. First order rate constants in 80 vol.% ethanol at 90.00° C a.

	k _{obs} (s ⁻¹)	^k rel	k _f	k _h
<u>5a</u>	4.07×10^{-6}	1		3.26×10^{-6}
<u>5b</u>	2.72×10^{-5}	7	2.72×10^{-6}	1.63×10^{-5}
<u>5c</u>	7.61×10^{-3}	1870	5.33×10^{-3}	5.33×10^{-4}

a) with 1.5 equiv. of triethylamine

The observed first order rate constants $(k_{\rm obs})$ for the tosylates $\underline{5}$ in 80 vol.% ethanol (Table) are the sum of the rate constants for fragmentation $(k_{\rm f})$, hydrolysis $(k_{\rm h})$ and resinification. Since the ratios of the latter rate constants are proportional to the known yields of the corresponding products, $k_{\rm f}$ and $k_{\rm h}$ can be calculated approximately (Table).

Surprisingly, hydrolysis of the oxime tosylates $\underline{5}$ competes with fragmentation. Furthermore both reactions are accelerated by electron releasing p-substituents, especially the latter reaction. This indicates a concerted fragmentation mechanism involving cleavage of the N-OTs and C(1)-C(2) bonds in the transition state $\underline{11}$ leading to the vinyl cations $\underline{10b}$ and $\underline{10c}$, respectively $\underline{^3}$. These reactions appear to be the first examples of vinyl cation formation by heterolytic fragmentation $\underline{^4}$.

$$H_3C$$
 $C = C$
 $C = N$
 OTS
 H_3C
 H_3C

The ready hydrolysis of the oxime tosylates 5 cannot be explained by a conventional mechanism involving nucleophilic addition of water to the C=N group 5. Rather, the increase of the hydrolysis rates k_h caused by electron releasing p-substituents (Table) indicates aryl-4-participation in the transition state with formation of phenonium ions 12 6. This is strikingly confirmed by the observation that 5a reacts ca. 75 times as fast as its analogue 13 in which the 2-phenyl group is absent. Addition of water to the strained C=N bond in 12 should lead to 14 which in turn could decompose to a cyclic ketone 6 and to imine (NH), a possible participant in the formation of the tarry material which is formed in the solvolysis of all the oxime tosylates 5.

REFERENCES

- 1) C.A. Grob & P. Wenk, Tetrahedron Letters
- 2) C.A. Grob & G. Cseh, Helv. chim. acta <u>47</u>, 194 (1964); C.A. Grob & H.R. Pfändler, ibid. <u>54</u>, 2060 (1971).
- 3) α -Aryl-substituted vinyl cations were first obtained by solvolysis of α -bromostyrenes 2 .
- 4) C.A. Grob & P.W. Schiess, Angew. Chemie, internat. Edit., 6, 1 (1967).
- 5) The oximes 5 (H for Ts) are not hydrolyzed under the reaction conditions.
- 6) Aryl-4-participation is usually considered to be unimportant, cf. B. Capon, Quarterly Reviews 18, 45 (1964).